Station 1:

Identify metal piece based on density.

|  |  |
| --- | --- |
| Substance | Density (g/mL) |
| Zinc (Solid) | 7.13 |
| Silver (Solid) | 10.50 |
| Lead (Solid) | 11.35 |

Identify: Suppose that 273 g of one of the substances listed above displaces 26 mL of water. What is the substance?

1. Zinc
2. Silver
3. Lead



Station 2:

Changes of States Cards

Instructions:

1. There are three picture cards. Place them face down on the table.
2. Each player picks a picture card.
3. Shuffle the other cards and spread them face up on the table.
4. Take turns top pick a card. If it describes your picture card, keep it. If not, you must give it to the player who needs it!
5. The game continues until all players have completed their sets.

Station 3:

Physical/Chemical Changes



**Station 4: ISTEP+ REVIEW**



SCAN ANSWERS ABOVE



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